

China Factory Wholesale Price Good Quality Hcds Si2cl6 Hexachlorodisilane

Basic Information

. Place of Origin: China . Brand Name: CMC COA · Certification: Si2cl6 Model Number: • Minimum Order Quantity: 1kg • Price: US \$40/kg · Packaging Details: Cylinder/Tank • Delivery Time: 15 days Payment Terms: L/C, T/T . Supply Ability: 50000kg/month



Product Specification

Product Name: Hexachlorodisilane

Model No.: Si2cl6 • Transport: By Sea 99.9% • Purity: 40L, 200L · Specification: Trademark: CMC • Origin: China 2812190091 HS Code: • Supply Ability: 100t/Year 7783-82-6 · CAS No.: Si2cl6 Formula: 7783-82-6 . EINECS:

Constituent: Industrial Pure Air
 Grade Standard: Industrial Grade
 Chemical Property: Poisonous Gases



More Images









Product Description

Product Description

Hexachlorodisilane (Si2Cl6) gas is a chemical compound composed of two silicon (Si) atoms and six chlorine (Cl) atoms. It is a colorless gas with a pungent odor and is primarily used in the field of organosilicon chemistry.

Here are some key points about hexachlorodisilane gas:

Chemical Structure: Hexachlorodisilane consists of two silicon atoms bonded to each other by a covalent bond, with each silicon atom being bonded to three chlorine atoms. The molecular formula is Si2Cl6. It is a highly reactive compound due to the presence of the silicon-chlorine bonds.

Synthesis: Hexachlorodisilane is typically synthesized by the reaction of silicon with chlorine gas. The reaction is highly exothermic and requires careful control of reaction conditions. It is often produced and used in situ due to its reactivity and tendency to decompose upon storage. Reactivity: Hexachlorodisilane is a versatile precursor in organosilicon chemistry. It can undergo various reactions to form a wide range of organosilicon compounds. It is commonly used as a source of silicon in the synthesis of silicon-based polymers, coatings, and other materials. Hexachlorodisilane can react with organic compounds containing functional groups like alcohols, amines, or thiols to introduce silicon atoms into the organic backbone.

Safety Considerations: Hexachlorodisilane is a reactive and corrosive compound that should be handled with caution. It can decompose spontaneously, releasing corrosive hydrogen chloride gas. It reacts vigorously with water, so proper safety measures, including the use of appropriate personal protective equipment and adherence to good laboratory practices, should be followed when working with hexachlorodisilane. Industrial Applications: Hexachlorodisilane has industrial applications in the production of organosilicon compounds and materials. It is used as a precursor in the synthesis of silicone polymers, resins, and elastomers. These materials find applications in various industries, including construction, electronics, automotive, and healthcare.

It's important to note that hexachlorodisilane is a specialized compound primarily used in research and industrial settings related to organosilicon chemistry. The average person is unlikely to encounter or work with this gas in their daily lives.

Basic Info.

Model NO.	Si2Cl6	Grade Standard	Electron Grade
Transport Package	Cylinder, Canister	Specification	40L, 200L
Trademark	СМС	Origin	Suzhou, China
HS Code	2812190091	Production Capacity	100t/Year

Specifications:

IUPAC nameHexachlorodisilaneOther namesDisilicon hexachloride

Identifiers

CAS No.: 13465-77-5 **EC No.:** 236-704-1

Properties

Molecular Formula:Si2Cl6Molar mass:268.88 g/molAppearance:Colorless liquid

Melting point: ≤20 °C

Boiling point: 144 °C (291 °F; 417 K)

Flash point: >93°C

Vopor density(Air=1): >1 Relative density(Water=1): 1.562

Sample Test:

Test Items	Units	Specifications	Test Result
Assay by GC	wt%	≥99.9	99.905
Li	ng/g	≤0.5	< 0.05
Na	ng/g	j ≤0.5	< 0.05
Mg	ng/g	յ ≤0.5	< 0.05
Al	ng/g	j≤1.0	0.35
K	ng/g	յ ≤0.5	0.08
Ca	ng/g	յ ≤0.5	0.16
Ti	ng/g	J≤1.0	0.18
Cr	ng/g	յ ≤0.5	< 0.05
Mn	ng/g	յ ≤0.5	< 0.05
Fe	ng/g	J≤1.0	0.48
Co	ng/g	J≤0.5	< 0.05

Detailed Photos









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Our products mainly include: H2, O2, N2, Ar, CO2, propane, acetylene, helium, laser mixed gas, SiH4, Sih2cl2, SiHCL3, SiCL4, NH3, CF4, NF3, SF6, HCL, N2O, doping mixed gas (TMB, PH3, B2H6) and other electronic gases.

CH3F H₂S F6+CI2 NH3 WF6 SiCl4 NH3 SiH4 Kr

C2 HCI+Ne 4MS C3F8 C3F8 **TEOS** CH4 PH₃ SF6

C4F8 SiH2 CF4

SiF4 **C3H8** CI2

DCE C3H6 BBr3

POCI3 **SO2** N2

BCI3 D2 CO2

CH2F2

SiHCI3

TMAI DMZn

DEZn

HF

AsH3

GeH4

C2H6

C2H4

B2H6

C2H2

H2Se

HBr

GeCl4

COS

Xe+NO

TMB+H2

He +As

Ge+Se

D+B

CO+NO

Ar+O2







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